

# Concentrations, sources, and influential factors of water-soluble ions of atmospheric particles in Dunhuang Mogao Grottoes, a world heritage site in China

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**Abstract:** Atmospheric particle pollution is one of the major factors leading to degradation of ancient wall paintings, particularly heritage sites in arid and semi-arid regions. However, current systematic research on the changes, sources, and influential factors of atmospheric particulate matter and its water-soluble ion concentrations is not sufficient. Thus, the major water-soluble ion concentrations, sources, and influential factors of atmospheric particles PM<sub>2.5</sub> and PM<sub>10</sub> (particulate matter with an aerodynamic equivalent diameter  $\leq 2.5$  and  $10.0 \mu\text{m}$ , respectively, in ambient air) were collected from Cave 16 and its ambient exterior environment in the Dunhuang Mogao Grottoes, China, between April 2015 and March 2016. Results showed that the concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> inside and outside the cave were the highest in March 2016 and the lowest in December 2015. The higher particle concentration from March to May was related to the frequent occurrence of sand and dust events, and the lower particle concentration from June to September was associated with good diffusion conditions, increased precipitation, and an established cave shelterbelt. The concentration of particulate matter inside the cave was affected by the concentration of particles in the air outside the cave.  $\text{Ca}^{2+}$ ,  $\text{NH}_4^+$ ,  $\text{Na}^+$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$  were the main components of the total ions of PM<sub>2.5</sub> and PM<sub>10</sub> both inside and outside the cave. The total ions inside the cave were frequently affected by the disturbance of tourists' activities during the peak tourist season from May to August. Under the influence of dust, the total concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$  in particles of different sizes inside and outside the cave increased, and the concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ , and  $\text{Ca}^{2+}$  decreased during precipitation period. Backward air mass trajectory analysis suggested that the pollutants were mainly from Xinjiang, China. The pollutant sources of air particulates are straw burning, secondary pollution sources, soil dust, dry spring rivers, and tourist activities.

**Keywords:** grotto temple; atmospheric particulate matter pollution; water-soluble ion; water and salt transport; heritage preventive conservation

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## 1 Introduction

Cultural heritage is the common wealth of humankind, and climate change, natural disasters, environmental pollution, and unreasonable management have caused serious damage to heritage conservation (Abbott, 2010; Li et al., 2013; Merello et al., 2014; Chatoutsidou et al., 2015; Vidal et al., 2019; Sakka et al., 2020; Ogura et al., 2021). Air particulates have a significant impact on global and regional climate change and environmental systems (Bouchlaghem et al., 2009; Guan et al., 2017; Yao et al., 2019), and damage cultural relics through coverage pollution and chemical corrosion (Pipal et al., 2014; Uring et al., 2019; Zhang et al., 2022). The deposition of atmospheric particulates on the surface of cultural relics change the color and texture of these relics, and affect their aesthetic value and visual effects (Brimblecombe, 1990; Natarajan et al., 2022), and certain chemicals in the dust will also affect murals and painted sculptures. The acidic components, such as sulfate, nitrate, and other small molecular organic acids, react with mineral pigments in the painted layers during the process of spreading and sedimentation. Long-term accumulation will cause corrosion and damage to cultural relics in caves (Hanapia and Din, 2012; Ogura et al., 2021), and the carbon and acid components will also react with  $\text{CaCO}_3$  to form a black or grey crust, which will further corrode the pigment layer (Ghedini et al., 2000; Sabbioni et al., 2003). The process of cleaning particles will cause further mechanical wear. Numerical simulations conducted by T  reault (2003) showed that for the preservation of cultural relics for 100 a, the maximum allowable concentration of  $\text{PM}_{2.5}$  should be limited to  $0.1 \mu\text{g}/\text{m}^3$ . Under the environmental conditions of  $\text{PM}_{2.5}$  of  $10.0 \mu\text{g}/\text{m}^3$ , the time of nondestructive preservation of cultural relics is only 1 a.

As early as the middle of the 19<sup>th</sup> century, the National Gallery of England had already begun to study the corrosive effects of the inside atmosphere on cultural relics (Broughton et al., 1875). Ghedini et al. (2011) analyzed the composition of water-soluble ions in atmospheric particulate matter surrounding the buildings of the Florence Cathedral. Singh and Sharma (2012) studied the seasonal changes and sources of  $\text{PM}_{10}$  in the Taj Mahal in India, and found that  $\text{PM}_{10}$  had an obvious seasonal change, and its source was mainly soil crust, vehicle, and industrial emissions. Chinese scientists conducted comprehensive monitoring and researches on inside and outside air quality in the Emperor Qin Shihuang's Mausoleum, Terra-cotta Warriors, Horses Museum, Exhibition Hall Museum, and the Jinsha Site Museum, and it was found that water-soluble components have caused long-term harm to cultural relics (Cao et al., 2005; Hu et al., 2009; Li et al., 2019; Deng et al., 2022).

Ancient murals are important components of cultural relics. The Dunhuang Mogao Grottoes ( $40^\circ 05' 17''\text{N}$ ,  $94^\circ 40' 16''\text{E}$ ; 1334 m a.s.l.), famous for their ancient murals, are situated at the western end of the Hexi Corridor of the Gansu Province, northwestern China. The Mogao Grottoes, which comprise of 492 caves,  $45 \times 10^3 \text{ m}^2$  of murals, and 2400 painted sculptures (Fan, 2000), was designated as a World Heritage Site by UNESCO (United Nations Educational, Scientific and Cultural Organization) in 1987 (Agnew, 2004). Because of its location on the southeastern edge of the Kumtag Desert and the eastern foot of the Mingsha Mountain, the dry climate provides unique natural conditions for the long-term preservation of the murals. The study area has a strong northwesterly and southwesterly wind, and a large amount of sand is transported to the caves, even into the caves, causing serious harm to the murals outside and inside the caves (Wang, 2018). Sand particles carried by the southeasterly wind with high velocity may contribute to the deterioration of the paintings on the eastern wall of the Mogao Grotto Cave 285 (Mikayama et al., 2015). The windblown sand causes the murals to undergo detachment and flaking (Ogura et al., 2021). Small air particles with a large specific surface area easily carry heavy metals, acidic oxides, organic pollutants, and microorganisms (Poulain et al., 2021). When the humidity is high, a shell-like coating readily forms on the surface of cultural relics. Small particles enter the gaps of the murals, speeding up the flaking of the murals. When visitors come into the caves, the air currents in the caves change (Wang et al., 2016), and the air particles in the caves will repeatedly settle and adsorb on the surface of the cultural relics, leading to aggravation of the mural coverage

and health of tourists and staff.

At present, the research on atmospheric particulate matter pollution in the Mogao Grottoes mainly focuses on microbes, elements, organic carbon, and water-soluble ions (Wang et al., 2011; He et al., 2020; Xu et al., 2020; Yang et al., 2021). This study addresses the current situation of atmospheric particulate matter pollution in the Mogao Grottoes based on an analysis of the temporal and spatial changes in particulate matter and its water-soluble ions, the characteristics of atmospheric particulate matter concentration, water-soluble ions under extreme conditions (precipitation and sandstorms), and changes in water-soluble ions. Moreover, through the analysis of the airflow backward trajectory, we speculated the source of atmospheric pollutants in the Mogao Grottoes in the dusty weather, and the changes in the concentrations of atmospheric particles and their water-soluble ions inside and outside the Mogao Grottoes. We systematically and comprehensively analyzed their sources and influential factors, based on the following questions: (1) what are the differences in particle changes inside and outside the caves? (2) what is the source of water-soluble ions in atmospheric particles? and (3) what are the effects of natural factors and visitors' activities? The results are expected to provide important information for the control and treatment of air pollution in the Mogao Grottoes, and provide a reference for building a monitoring and early warning system for cultural relic protection and tourism opening decision-making.

## 2 Materials and methods

### 2.1 Study area

The Mogao Grottoes are 25 km from the Dunhuang City, surrounded by the Gobi area and river systems. The sampled sites were located in front of Cave 72 (outside the cave) and in Cave 16 (inside the open cave), which is 1 km west of the Daquan River, and the Kumtag Desert is located to the east (Fig. 1). The climate is extremely arid, with average annual precipitation of 39 mm, of which more than 59% is concentrated mainly from June to August. The annual relative humidity is 31%, the annual average temperature is 11.2°C, and the annual evaporation is about 4348 mm (Liu et al., 2020).



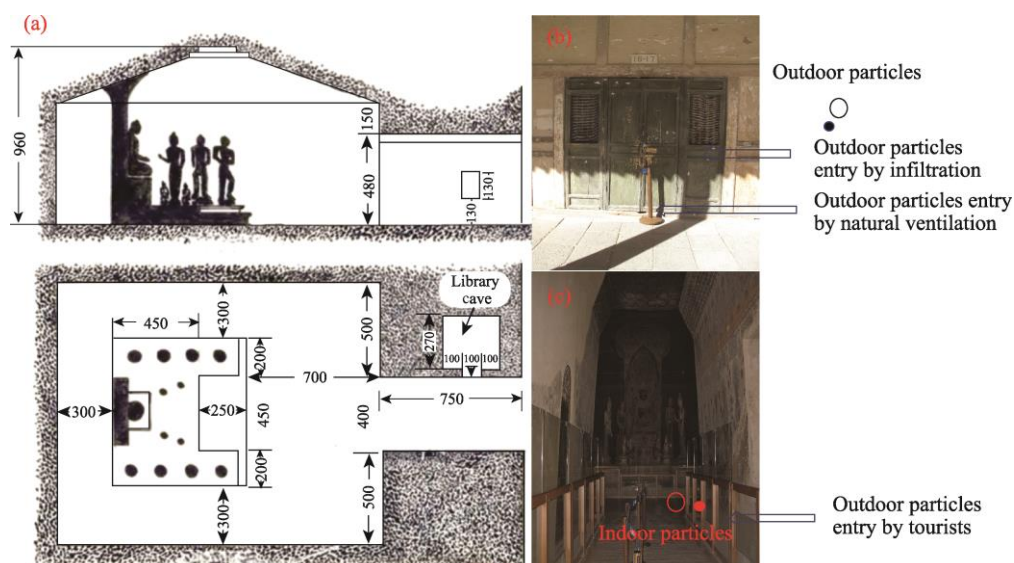
**Fig. 1** Location of the sampling sites and air particulate pollution at the Dunhuang Mogao Grottoes. (a), aerial photo of the study area; (b), sandstorm pollution outside the cave; (c), dust reduction on the surface of sculptures inside the cave; (d), orthophoto of the Dunhuang Mogao Grottoes; (e), air particulate matter pollution inside the cave; (f), sampling point outside the cave; (g), sampling point inside the cave.

## 2.2 Plane section and air exchange rate of the cave

During the sampling period, the total number of tourists in the Mogao Grottoes was 1,147,735, and the monthly average number of tourists from April to December 2015 was 36,920, 66,065, 101,503, 216,793, 283,197, 168,084, 202,511, 22,699, and 8275, respectively. From January to March 2016, the monthly average number of tourists was 7605, 17,275, and 16,808, respectively.

When there are no tourists, the air exchange rate of small and medium-sized caves is generally less than 10 min. When there are tourists, the air exchange rate inside and outside the cave is correspondingly doubled, and the impact of tourists on the air exchange rate inside and outside the cave is obvious (Zhang et al., 2009).

The width of the door of Cave 16 is 4.0 m, the distance between cave door and tunnel is 7.5 m, and the distance between the entrance of tunnel and sampling instrument is 14.5 m. The cave door has screen windows, and the cave has wooden floors. There are three main ways for particles to enter the cave, i.e., infiltration, natural ventilation, and tourists (Fig. 2).



**Fig. 2** Pathways of particles entering the cave. (a), hand drawing of Cave 16 of the Mogao Grottoes, 2004 (the measurement unit in the figure is cm); (b), wooden cave door; (c), inside the cave.

## 2.3 Air particulate matter data

PM<sub>2.5</sub> and PM<sub>10</sub> samples were collected on quartz fiber filter papers (47 mm, Whatman, USA) by two parallel medium-volume air samplers (Dichotomous Partisol-Plus Model 2025, Thermo Scientific, USA) at each sampling site, at the flow rates of 15.00 and 1.67 L/min for 24 h, respectively, with separate samplings during dust and precipitation. At least 316 replicate samples were obtained from two sampling sites. To remove carbonaceous contaminants, we pre-heated the quartz filters at 500°C for 4 h in a muffle furnace before sampling. Before and after sampling, we equilibrated the filters for 24 h in a balance dish at a temperature of 20°C (±1)°C and a relative humidity of 50% (±5%). Before and after sampling, we weighed the weight of the filter membrane with electronic balance (AUY220, Shimadzu, Japan) three times. After sampling, we stored the filter membrane in a refrigerator below 0°C prior to weighing and water-soluble ion analysis.

## 2.4 Meteorological data

Air temperature, relative humidity, rainfall, wind speed, and wind direction were obtained from the weather station in front of Cave 72. These data in Gobi area were provided by the monitoring and early warning system platform of the Cave Monitoring Center of Dunhuang Academy.



## 2.5 Water-soluble ion sample preparation and detection

Major cations ( $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ , and  $\text{NH}_4^+$ ) and major anions ( $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$ , and  $\text{SO}_4^{2-}$ ) of selected air particulate matter samples were measured using an ion chromatograph (Dionex-6000, Dionex, USA) (Li et al., 2016).

## 2.6 Quality assurance and control

All analytical procedures complied with strict quality assurance and control. The correlation coefficients of the standard working curves for the 9 water-soluble ions were all greater than 0.9990. All water-soluble ion concentrations had already been corrected by eliminating field blank values to eliminate the error caused by gas adsorption artifacts. The detection limit was 1 ng/mL, and the standard deviation was less than 5%. The determination limits for the ions were as follows:  $\text{F}^-$  (0.002  $\mu\text{g}/\text{m}^3$ ),  $\text{Cl}^-$  (0.001  $\mu\text{g}/\text{m}^3$ ),  $\text{NO}_3^-$  (0.004  $\mu\text{g}/\text{m}^3$ ),  $\text{SO}_4^{2-}$  (0.004  $\mu\text{g}/\text{m}^3$ ),  $\text{Ca}^{2+}$  (0.003  $\mu\text{g}/\text{m}^3$ ),  $\text{Mg}^{2+}$  (0.003  $\mu\text{g}/\text{m}^3$ ),  $\text{Na}^+$  (0.003  $\mu\text{g}/\text{m}^3$ ),  $\text{K}^+$  (0.006  $\mu\text{g}/\text{m}^3$ ), and  $\text{NH}_4^+$  (0.003  $\mu\text{g}/\text{m}^3$ ).

## 2.7 Statistical analysis

Statistical analysis was performed using SPSS v.22.0, and the graphs were drawn using Origin v.16.0. The Pearson method was used to analyze the correlations among air temperature, air pressure, relative humidity, wind speed, precipitation, number of tourists, and concentration of atmospheric particulate matter inside and outside the cave.

# 3 Results

## 3.1 Air particulate matter concentration

As shown in Figure 3, the monthly average concentrations of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  at the two monitoring points showed a similar trend, with a large monthly change. The concentrations of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave were both at the highest in March 2016 (339.4677, 421.6259, 708.2016, and 754.6400  $\mu\text{g}/\text{m}^3$ ) and the lowest in December 2015 (15.4841, 23.6697, 21.4576, and 27.7507  $\mu\text{g}/\text{m}^3$ ). Low concentrations of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  (variation ranges of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside the cave were 15.4841–49.5065 and 23.6697–74.0431  $\mu\text{g}/\text{m}^3$ , respectively; variation ranges of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  outside the cave were 21.4576–113.8568 and 27.7507–132.8765  $\mu\text{g}/\text{m}^3$ , respectively) were mainly concentrated from November 2015 to February 2016. The concentrations of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside the cave from June to September were different from those outside the cave (variation ranges of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside the cave were 49.8075–60.6085 and 70.0128–82.1617  $\mu\text{g}/\text{m}^3$ , respectively; variation ranges of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  outside the cave were 58.7113–88.8595 and 69.1679–118.4600  $\mu\text{g}/\text{m}^3$ , respectively).

$\text{PM}_{2.5}/\text{PM}_{10}$  ratios inside and outside the cave were both less than 1, indicating that the atmospheric particulate matter pollution of the Mogao Grottoes mainly came from coarse

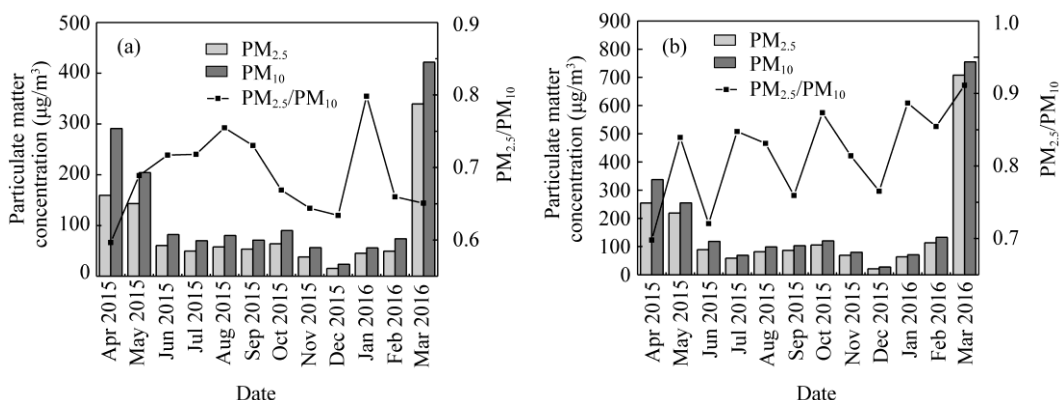


Fig. 3 Monthly variation of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  concentrations inside (a) and outside (b) the cave

particles with a diameter greater than  $PM_{10}$ . It is also clear that  $PM_{2.5}/PM_{10}$  increased from the beginning to 15 August, and then decreased with a jump in January 2016 inside the cave, as well as a fluctuated increase outside the cave.

Correlation analysis revealed that  $PM_{2.5}$  mass concentration inside the cave was significantly and positively correlated with that outside the cave ( $r=0.867$ ,  $P<0.01$ ), and the  $PM_{10}$  mass concentration was also significantly and positively correlated ( $r=0.903$ ,  $P<0.01$ ). This showed that the concentration of particulate matter in the cave was mainly affected by the concentration of particulate matter in the atmosphere outside the cave.

Table 1 showed the correlation analysis of  $PM_{2.5}$  and  $PM_{10}$  concentrations inside and outside the cave with various meteorological elements and number of tourists. The concentrations of  $PM_{2.5}$  and  $PM_{10}$  at the two sampling points were negatively correlated with air temperature, relative humidity, and precipitation. During the sampling period, the temperature outside and inside the cave varied from  $-16.66^{\circ}\text{C}$  to  $40.71^{\circ}\text{C}$  and  $-3.36^{\circ}\text{C}$  to  $23.00^{\circ}\text{C}$ , respectively. The concentration of particulate matter inside the cave had a higher correlation with temperature than that of outside the cave. Temperature difference promotes air exchange. Annual precipitation of the Mogao Grottoes is 39 mm, concentrated in May–September. Concentrations of  $PM_{2.5}$  and  $PM_{10}$  at the two sampling points were positively correlated with air pressure, wind speed, and number of tourists. The concentration of particulate matter in the cave had a strong correlation with number of tourists and wind speed, indicating that the number of tourists contributed more to the air pollution in the cave.

**Table 1** Correlation coefficients of  $PM_{2.5}$  and  $PM_{10}$  inside and outside the cave with meteorological factors and number of tourists

| Sampling point     | Air temperature | Relative humidity | Air pressure | Wind speed | Precipitation | Number of tourists |
|--------------------|-----------------|-------------------|--------------|------------|---------------|--------------------|
| $PM_{2.5}$ outside | -0.164          | -0.046            | 0.197        | 0.217      | -0.382        | 0.586 <sup>*</sup> |
| $PM_{10}$ outside  | -0.042          | -0.123            | 0.107        | 0.299      | -0.395        | 0.489              |
| $PM_{2.5}$ inside  | -0.191          | -0.166            | 0.081        | 0.345      | -0.271        | 0.615 <sup>*</sup> |
| $PM_{10}$ inside   | -0.160          | -0.192            | 0.083        | 0.377      | -0.281        | 0.598 <sup>*</sup> |

Note: \*,  $P<0.05$  level.

### 3.2 Variation of water-soluble ions inside and outside the cave

#### 3.2.1 Concentration of water-soluble ions

As shown in Table 2, the total concentrations of water-soluble ions in  $PM_{2.5}$  and  $PM_{10}$  inside the cave were 11.0106 ( $\pm 5.4744$ ) and 22.3576 ( $\pm 10.5183$ )  $\mu\text{g}/\text{m}^3$ , respectively, and the ion concentration was ranked as  $\text{Ca}^{2+}>\text{NH}_4^+>\text{Cl}^->\text{Na}^+>\text{SO}_4^{2-}>\text{K}^+>\text{NO}_3^->\text{Mg}^{2+}>\text{F}^-$ . Those outside the

**Table 2** Concentrations of water-soluble ions of  $PM_{2.5}$  and  $PM_{10}$  inside and outside the cave

| Water-soluble ion                | Inside the cave                         |  |                     | Outside the cave                        |  |                     |
|----------------------------------|---|--|---------------------|---|--|---------------------|
|                                  | $PM_{2.5}$ ( $\mu\text{g}/\text{m}^3$ ) | $PM_{10}$ ( $\mu\text{g}/\text{m}^3$ ) | $PM_{2.5}/PM_{10}$  | $PM_{2.5}$ ( $\mu\text{g}/\text{m}^3$ ) | $PM_{10}$ ( $\mu\text{g}/\text{m}^3$ ) | $PM_{2.5}/PM_{10}$  |
| $\text{F}^-$                     | 0.0244 $\pm$ 0.0222                     | 0.0406 $\pm$ 0.0292                    | 0.5651 $\pm$ 0.1571 | 0.0121 $\pm$ 0.0065                     | 0.0236 $\pm$ 0.0146                    | 0.5488 $\pm$ 0.1378 |
| $\text{Cl}^-$                    | 1.8148 $\pm$ 1.1201                     | 3.2967 $\pm$ 1.9301                    | 0.5640 $\pm$ 0.1230 | 2.1414 $\pm$ 1.8429                     | 3.6763 $\pm$ 2.5832                    | 0.6117 $\pm$ 0.1379 |
| $\text{SO}_4^{2-}$               | 1.0807 $\pm$ 1.1201                     | 2.0888 $\pm$ 1.7851                    | 0.4883 $\pm$ 0.1424 | 1.5293 $\pm$ 2.1851                     | 2.6484 $\pm$ 2.7977                    | 0.5498 $\pm$ 0.1490 |
| $\text{NO}_3^-$                  | 0.4009 $\pm$ 0.3349                     | 0.7279 $\pm$ 0.6338                    | 0.5498 $\pm$ 0.1041 | 0.5230 $\pm$ 0.4101                     | 0.8858 $\pm$ 0.7251                    | 0.6204 $\pm$ 0.1027 |
| $\text{Na}^+$                    | 1.1287 $\pm$ 0.8054                     | 2.6107 $\pm$ 1.7141                    | 0.4496 $\pm$ 0.1555 | 1.1014 $\pm$ 0.7051                     | 3.3736 $\pm$ 3.8020                    | 0.3935 $\pm$ 0.1804 |
| $\text{NH}_4^+$                  | 1.8427 $\pm$ 1.2874                     | 2.9178 $\pm$ 1.9891                    | 0.6217 $\pm$ 0.1770 | 1.6681 $\pm$ 1.1033                     | 2.7246 $\pm$ 1.4897                    | 0.5959 $\pm$ 0.1633 |
| $\text{K}^+$                     | 0.4718 $\pm$ 0.2525                     | 0.9396 $\pm$ 0.4702                    | 0.5023 $\pm$ 0.1133 | 0.4510 $\pm$ 0.2507                     | 0.9211 $\pm$ 0.5518                    | 0.4909 $\pm$ 0.1009 |
| $\text{Mg}^{2+}$                 | 0.2946 $\pm$ 0.2499                     | 0.5676 $\pm$ 0.3226                    | 0.4759 $\pm$ 0.1846 | 0.2397 $\pm$ 0.1874                     | 0.5367 $\pm$ 0.3529                    | 0.4288 $\pm$ 0.1457 |
| $\text{Ca}^{2+}$                 | 3.9520 $\pm$ 2.8727                     | 9.1679 $\pm$ 4.2838                    | 0.4230 $\pm$ 0.1907 | 4.0340 $\pm$ 3.3046                     | 10.5977 $\pm$ 5.7177                   | 0.3651 $\pm$ 0.2167 |
| TIC                              | 11.0106 $\pm$ 5.4744                    | 22.3576 $\pm$ 10.5183                  | -                   | 11.7001 $\pm$ 7.1713                    | 25.3878 $\pm$ 15.7937                  | -                   |
| $\text{NO}_3^-/\text{SO}_4^{2-}$ | 0.4206 $\pm$ 0.2183                     | 0.3471 $\pm$ 0.1319                    | -                   | 0.4420 $\pm$ 0.2554                     | 0.3617 $\pm$ 0.1524                    | -                   |

Note: TIC, total ion concentration. - means no value. Mean  $\pm$  SD.

cave were 11.7001 ( $\pm 7.1713$ ) and 25.3878 ( $\pm 15.7937$ )  $\mu\text{g}/\text{m}^3$ , respectively, and ranked as  $\text{Ca}^{2+} > \text{Cl}^- > \text{NH}_4^+ > \text{SO}_4^{2-} > \text{Na}^+ > \text{NO}_3^- > \text{K}^+ > \text{Mg}^{2+} > \text{F}^-$ . The result showed that  $\text{Ca}^{2+}$ ,  $\text{NH}_4^+$ ,  $\text{Na}^+$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$  were the main components of the total water-soluble ions of atmospheric air particulate matter both inside and outside the cave. The ratios of  $\text{PM}_{2.5}/\text{PM}_{10}$  inside and outside the cave were 0.42–0.62 and 0.37–0.62, respectively, showing that 9 kinds of water-soluble ions were different in the degree of enrichment in particles of different sizes. The concentration of  $\text{NH}_4^+$  in the fine particles inside the cave was relatively high, and the concentrations of  $\text{Cl}^-$ ,  $\text{NO}_3^-$ , and  $\text{NH}_4^+$  outside the cave were relatively high in the fine particulate matter.

It is clear that the ratio of each ion to total ion content was basically not changed, indicating that the composition of each ion was relatively stable, either inside or outside the cave (Fig. 4). The concentrations of  $\text{Cl}^-$  accounted for 16.48% and 14.76%, and 18.30% and 14.40%, of total water-soluble ions in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave, respectively. The concentrations of  $\text{NH}_4^+$  accounted for 16.74% and 13.05%, and 14.26% and 10.73%, respectively, while the concentrations of  $\text{Ca}^{2+}$  accounted for 35.89% and 41.01%, and 34.48% and 41.74%, respectively.

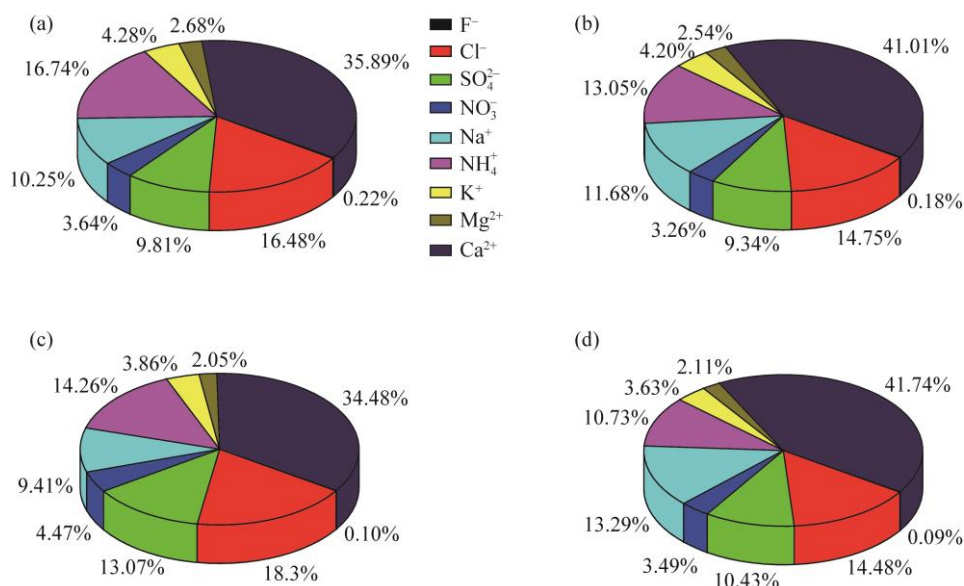
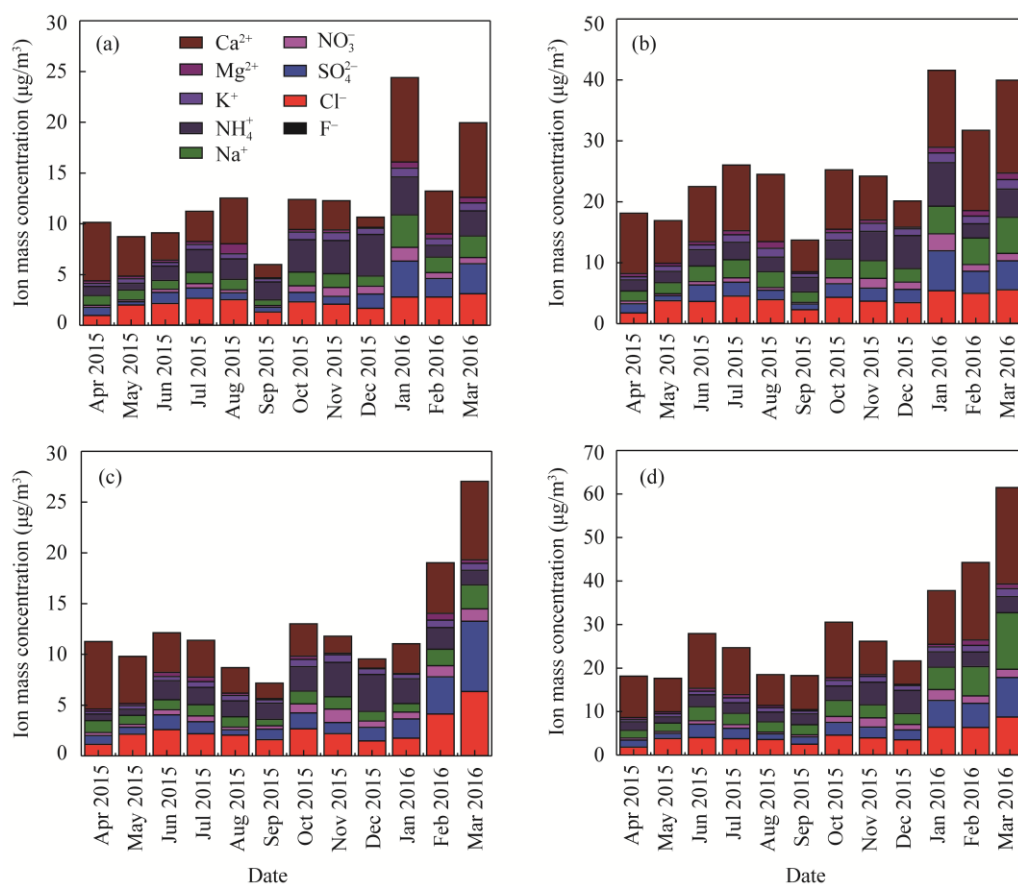


Fig. 4 Proportion of water-soluble ions in  $\text{PM}_{2.5}$  (a) and  $\text{PM}_{10}$  (b) inside and outside (c and d) the cave

### 3.2.2 Monthly change of water-soluble ions

Figure 5 shows that the total ion concentration in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  in the cave, in general, increased in the year of measurement and fall to the lowest in September, and increased to peak or secondary peak in January or March. The concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$  increased from January to March 2016. The concentration of  $\text{F}^-$  was higher in July–August 2015, the concentration of  $\text{NO}_3^-$  was higher in November 2015–January 2016, and the concentration of  $\text{NH}_4^+$  was higher in October 2015–January 2016.

Total ion concentration in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  outside the cave decreased from June 2015 to October, and then increased until January 2016, and the concentration of water-soluble ions showed an increasing trend from January to March 2016. The monthly distributions of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$  outside the cave were similar to those inside the cave.  $\text{F}^-$  changed from January to March 2016,  $\text{NH}_4^+$  concentration is high from October 2015 to March 2016. Both inside and outside of the cave, the ion mass loss is the lowest in September and the highest is in January or March.



**Fig. 5** Monthly average concentrations of water-soluble ions in PM<sub>2.5</sub> (a) and PM<sub>10</sub> (b) inside and outside (c and d) the cave

### 3.3 Air particulate and water-soluble ions in particles of different sizes inside and outside the cave in extreme conditions

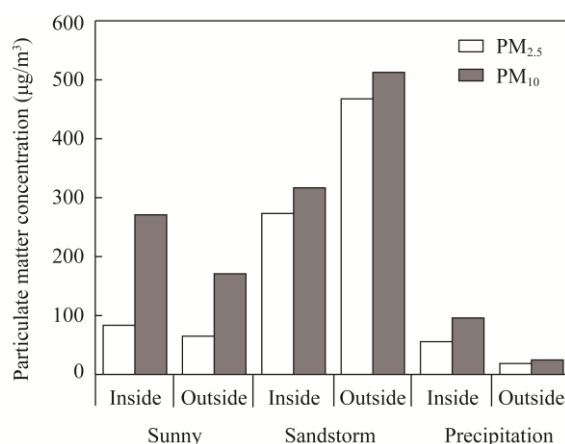
#### 3.3.1 Influence of sandstorm and precipitation on atmospheric particulate matter

A sandstorm occurred in Dunhuang from 01:00 (LST) to 19:40 on 29 April, 2015, with a wind speed of 7.83 m/s at the top of the cave, and reached a maximum instantaneous wind speed of 12.72 m/s at 15:40. During the sandstorm, the wind direction varied from 0.0 ° to 355.2 °, and the main wind direction was NNE (north-north-east), which is conducive to the movement of pollutants into the cave. There was continuous precipitation from 17:30 to 21:10 on 29 April, and the cumulative precipitation reached 2.3 mm, accounting for 17.04% of the total precipitation in 2015, and the relative humidity reached 84.56%. Therefore, samples were collected to analyze the effects of special weather on the concentration of atmospheric particulates and water-soluble ions at the two sampling points in the Mogao Grottoes.

Figure 6 shows that during sunny weather, the concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> inside and outside the cave were 83.3333 and 270.8333, and 64.8148 and 170.8333 μg/m<sup>3</sup>, and the concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> outside the cave were 0.78 and 0.63 times of the corresponding particle size concentration inside the cave. During precipitation period, the concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> inside and outside of the cave were 55.5556 and 95.8333, and 18.5185 and 24.5370 μg/m<sup>3</sup>, and the concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> outside the cave was 0.33 and 0.26 times of that inside the cave, respectively. During sandstorm, the concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> inside and outside the cave were 287.6157, 367.2535, 474.5370, and 547.9167 μg/m<sup>3</sup>, which were 3.45, 1.36, 7.32 and 3.21 times of those in the sunny weather, 5.18, 3.83, 25.63, and



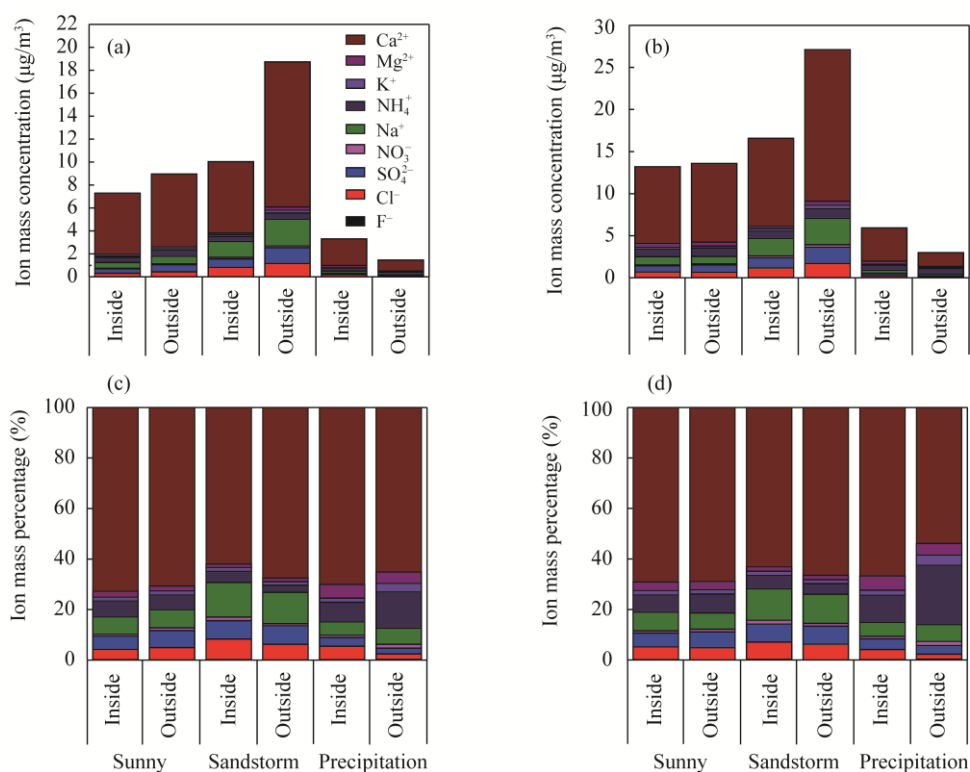
22.33 times of those during precipitation period. And the concentrations of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  outside the cave were 1.65 and 1.49 times of the corresponding particle size concentration inside the cave. Sandstorm and precipitation affected the concentration of atmospheric particulates rapidly at the two monitoring points, especially the concentration of pollutants outside the cave.



**Fig. 6** Concentration of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave of the Mogao Grottoes under sunny weather, sandstorm weather, and precipitation

### 3.3.2 Influence of sandstorm and precipitation on water-soluble ions

Water-soluble ion concentration in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  were different inside and outside the cave (Fig. 7). On sunny days, the total concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$  in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$



**Fig. 7** Concentrations of water-soluble ions in  $\text{PM}_{2.5}$  (a) and  $\text{PM}_{10}$  (b), relative proportion of soluble ions in  $\text{PM}_{2.5}$  (c) and  $\text{PM}_{10}$  (d) inside and outside the cave of the Mogao Grottoes under sunny weather, sandstorm weather, and precipitation

inside and outside the cave were 6.9426, 12.3699, 8.5362, and 12.7496  $\mu\text{g}/\text{m}^3$ , which accounted for 95.24%, 93.80%, 95.28%, and 93.88% of the total water-soluble ions, respectively. The total mass concentrations of other ions ( $\text{F}^-$ ,  $\text{NO}_3^-$ ,  $\text{K}^+$  and  $\text{Mg}^{2+}$ ) were 0.3469, 0.8180, 0.4225, and 0.8311  $\mu\text{g}/\text{m}^3$ , accounting for only 4.76%, 6.20%, 4.72%, and 6.12%, respectively.

During the sandstorm, the total concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$  in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave were 9.5812, 15.7339, 18.0314, and 25.8900  $\mu\text{g}/\text{m}^3$ , accounting for 95.51%, 94.91%, 96.28%, and 95.51% of the total water-soluble ions, respectively. The total mass concentrations of other ions ( $\text{F}^-$ ,  $\text{NO}_3^-$ ,  $\text{K}^+$ , and  $\text{Mg}^{2+}$ ) were 0.4502, 0.8434, 0.6971 and 1.2170  $\mu\text{g}/\text{m}^3$ , accounting for only 4.49%, 5.09%, 3.72%, and 4.49%, respectively.

During the precipitation period, the total concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$  in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave were 3.0384, 5.4094, 1.3198, and 2.6671  $\mu\text{g}/\text{m}^3$ , the proportions to the total water-soluble ions were 91.89%, 91.25%, 90.31%, and 89.39%, and the total mass concentrations of other ions ( $\text{F}^-$ ,  $\text{NO}_3^-$ ,  $\text{K}^+$ , and  $\text{Mg}^{2+}$ ) were 0.2681, 0.5188, 0.1416, and 0.3166  $\mu\text{g}/\text{m}^3$ , accounting for only 8.11%, 8.75%, 9.69%, and 10.61%, respectively.

Further analysis of the data showed that under the influence of dust weather, the concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ , and  $\text{Na}^+$  were 1.72–2.75, 1.63–2.29, 1.54–2.41, and 2.12–3.62 times of those on sunny days, and the proportion increased significantly. Concentrations of  $\text{F}^-$ ,  $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$  changed little. While the proportions of  $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$  decreased, the proportion of  $\text{F}^-$  in the cave increased, and the proportion outside the cave decreased.  $\text{Cl}^-$  and  $\text{Na}^+$  varied the most. The concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ , and  $\text{Ca}^{2+}$  during the sandstorm period were 4.65–36.51, 4.73–41.91, 6.30–24.87, and 2.65–13.28 times of those during precipitation period. The proportions of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ , and  $\text{Na}^+$  decreased significantly, and  $\text{Ca}^{2+}$  proportion increased in the cave and decreased outside the cave.

### 3.3.3 Backward trajectory of air flow

To identify potential transport pathways and possible source regions of the aerosols over the Mogao Grottoes, we used Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model ([http://ready.arl.noaa.gov/HYSPLIT\\_traj.php](http://ready.arl.noaa.gov/HYSPLIT_traj.php)) to simulate the transport of sand and dust, and calculated a 24-h backward trajectory (Fig. 8). Heights of 500, 1000, and 1500 m were selected, and the back calculation time was 17:00 (UTC) on 28 April and 01:00 (LST) on 29 April, 2015. Combined with the ion concentrations listed in Figure 7, air masses arriving at the sampling site were classified into two types: (1) air masses that arrived on 29 April, which originated from or were transported through the arid regions of Hami City, Xinjiang Uygur Autonomous Region of China, through the Taklimakan Desert; and (2) through the Kumtag Desert to arrive at the sampling site, which may have resulted in the input of desert dust and some anthropogenic pollution. Thus, the concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$  were very high.

### 3.4 Acidity and alkalinity of atmospheric particles

Acidity of atmospheric particles can be determined by analyzing the balance of the main anion and cation charges in the particles (Zhou et al., 2016):

$$\text{AE} = c(\text{Cl}^-)/35.5 + c(\text{NO}_3^-)/62 + c(\text{SO}_4^{2-})/48 + c(\text{F}^-)/19, \quad (1)$$

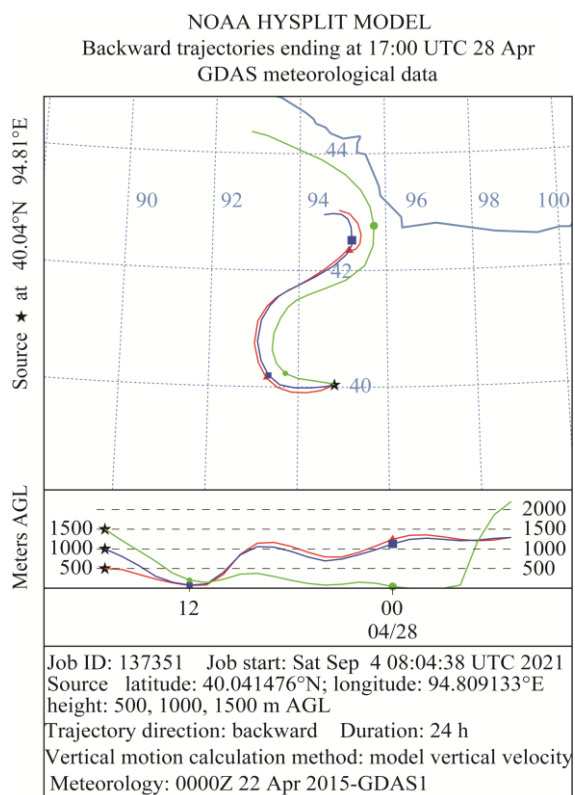
$$\text{CE} = c(\text{NH}_4^+)/18 + c(\text{Mg}^{2+})/12 + c(\text{K}^+)/39 + c(\text{Ca}^{2+})/20 + c(\text{Na}^+)/23, \quad (2)$$

where  $c$  is the concentration of each ion ( $\mu\text{g}/\text{m}^3$ ); AE is the anion equivalent ( $\mu\text{eq}/\text{m}^3$ ); and CE is the cation equivalent ( $\mu\text{eq}/\text{m}^3$ ).

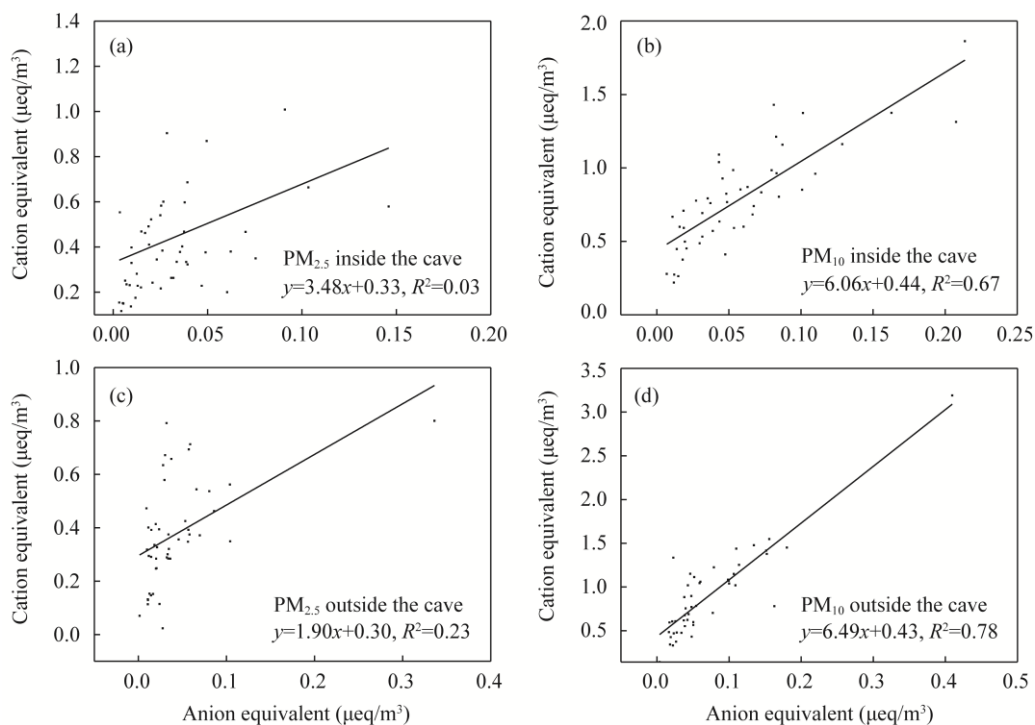
Figure 9 shows the fitting lines for equivalent values of cations to those of anions in  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$ , the slopes of the anion and cation balance of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave were greater than 1, i.e., 3.48, 6.06, 1.90, and 6.49, respectively. The results showed anion deficiency and alkalinity in the Mogao Grottoes.

### 3.5 Source analysis of water-soluble ions

The results of principal component analysis and orthogonal rotation factor load matrix are shown in Tables 3 and 4. As shown in Table 3,  $\text{PM}_{2.5}$  in the cave contributed 76.22%. Factor 1 was loaded with  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$ , and  $\text{SO}_4^{2-}$  and accounted for 40.73% of the total variance.



**Fig. 8** Twenty-four-hour backward trajectory on 29 April, 2015. NOAA, National Oceanic and Atmospheric Administration; HYSPLIT, Hybrid Single-Particle Lagrangian Integrated Trajectory. GDAS, Global Data Assimilation System AGL, above ground level.



**Fig. 9** Ion balance of  $\text{PM}_{2.5}$  (a and c) and  $\text{PM}_{10}$  (b and d) inside and outside the cave of the Mogao Grottoes

Factor 2 was dominated by  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$ . Factor 3 was dominated by  $\text{F}^-$ . In the cave (Table 4), the contribution rate of the three factors was 84.57% to the  $\text{PM}_{10}$ . The predominance of  $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{NO}_3^-$ , and  $\text{SO}_4^{2-}$  in factor 1 accounted for 33.89%.  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{F}^-$ , and  $\text{Cl}^-$  in factor 2 had significant effects. Factor 3 had significant effects on  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$ . In  $\text{PM}_{2.5}$  outside the cave (Table 3), the contribution rate of the three factors was 82.90%. Factor 1 dominated by  $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$ , and  $\text{SO}_4^{2-}$  had significant effects and accounted for 33.89%. In factor 2,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$  had significant effects. In factor 3,  $\text{F}^-$  has a significant effect. In  $\text{PM}_{10}$  outside the cave (Table 4), the contribution rate of the three factors was 88.28%. In Factor 1,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$  had significant effects and accounted for 51.27%. In Factor 2,  $\text{NH}_4^+$  and  $\text{NO}_3^-$  had significant effects.  $\text{F}^-$  in factor 3 had significant effects.

**Table 3** Varimax rotated factor loading matrix for water-soluble ions in  $\text{PM}_{2.5}$  inside and outside the cave

| Content               | Load (inside the cave) |                |                | Load (outside the cave) |                |                |
|-----------------------|------------------------|----------------|----------------|-------------------------|----------------|----------------|
|                       | F <sub>1</sub>         | F <sub>2</sub> | F <sub>3</sub> | F <sub>1</sub>          | F <sub>2</sub> | F <sub>3</sub> |
| $\text{Na}^+$         | 0.805                  | 0.287          | -0.348         | 0.329                   | 0.896          | 0.094          |
| $\text{NH}_4^+$       | 0.563                  | -0.305         | -0.472         | 0.070                   | -0.133         | 0.945          |
| $\text{K}^+$          | 0.865                  | 0.008          | 0.010          | 0.382                   | 0.606          | 0.639          |
| $\text{Mg}^{2+}$      | -0.045                 | 0.866          | 0.303          | 0.361                   | 0.605          | 0.149          |
| $\text{Ca}^{2+}$      | 0.144                  | 0.962          | -0.028         | 0.114                   | 0.880          | -0.359         |
| $\text{F}^-$          | 0.113                  | 0.306          | 0.764          | 0.823                   | 0.118          | 0.153          |
| $\text{Cl}^-$         | 0.705                  | 0.283          | 0.260          | 0.771                   | 0.498          | 0.060          |
| $\text{NO}_3^-$       | 0.894                  | -0.084         | 0.221          | 0.794                   | 0.238          | 0.407          |
| $\text{SO}_4^{2-}$    | 0.787                  | 0.001          | 0.157          | 0.863                   | 0.315          | -0.101         |
| Eigenvalue            | 3.665                  | 2.032          | 1.162          | 3.050                   | 2.746          | 1.665          |
| Contribution rate (%) | 40.730                 | 22.580         | 12.910         | 33.890                  | 30.520         | 18.500         |

**Table 4** Varimax rotated factor loading matrix for water-soluble ions in  $\text{PM}_{10}$  inside and outside the cave

| Content               | Load (inside the cave) |                |                | Load (outside the cave) |                |                |
|-----------------------|------------------------|----------------|----------------|-------------------------|----------------|----------------|
|                       | F <sub>1</sub>         | F <sub>2</sub> | F <sub>3</sub> | F <sub>1</sub>          | F <sub>2</sub> | F <sub>3</sub> |
| $\text{Na}^+$         | 0.511                  | 0.727          | 0.135          | 0.878                   | 0.292          | 0.183          |
| $\text{NH}_4^+$       | 0.929                  | -0.057         | 0.020          | 0.071                   | 0.846          | 0.415          |
| $\text{K}^+$          | 0.673                  | 0.603          | 0.146          | 0.729                   | 0.536          | 0.224          |
| $\text{Mg}^{2+}$      | 0.028                  | 0.046          | 0.989          | 0.835                   | 0.131          | 0.379          |
| $\text{Ca}^{2+}$      | 0.060                  | 0.350          | 0.911          | 0.902                   | 0.009          | 0.268          |
| $\text{F}^-$          | -0.034                 | 0.753          | 0.219          | 0.292                   | 0.347          | 0.856          |
| $\text{Cl}^-$         | 0.397                  | 0.830          | 0.129          | 0.853                   | 0.435          | 0.045          |
| $\text{NO}_3^-$       | 0.868                  | 0.319          | 0.012          | 0.485                   | 0.743          | 0.096          |
| $\text{SO}_4^{2-}$    | 0.745                  | 0.519          | 0.032          | 0.864                   | 0.378          | 0.077          |
| Eigenvalue            | 3.050                  | 2.647          | 1.914          | 4.614                   | 2.110          | 1.222          |
| Contribution rate (%) | 33.890                 | 29.420         | 21.260         | 51.270                  | 23.440         | 13.580         |

## 4 Discussion

### 4.1 Effects of climatic factors and visitors

In addition to pollution sources, concentration of atmospheric particulate matter is affected by meteorological elements such as temperature, air pressure, wind speed, and precipitation (Guan et al., 2019; Vidal et al., 2019). Precipitation is considered to be a major factor in the long-term

preservation of murals (Wang et al., 2018). When the relative humidity exceeds 75%, it will increase the frequency of mural cracks and also influences the process of material transformation (Ogura et al., 2021). In Figure 3, difference between concentrations of  $PM_{2.5}$  inside and outside the cave in our study (March) was much higher than that reported by Li et al. (2014) in the Pottery Depot of the Terra-cotta Warriors and Horses Museum of Qin Shihuang (inside,  $76.1 \mu\text{g}/\text{m}^3$ ; outside,  $153.9 \mu\text{g}/\text{m}^3$ ), and by Deng et al. (2022) in the Jinsha Site Museum (inside,  $33.3 \mu\text{g}/\text{m}^3$ ; outside,  $39.4 \mu\text{g}/\text{m}^3$ ), the difference of the result can be attributed to the windy and sandy weather in the arid environment of northern China. The cold air activity is more frequent and likely to cause sand and dust in the spring (from March to May) in Dunhuang. Mingsha Mountain that provides a sufficient sand source for windy weather is 1 km from the top of the Mogao Grottoes, although a shelterbelt locates directly in front of the Mogao Grottoes and provides protection against the sand and dust processes of the Mogao Grottoes (Xu et al., 2020).

The concentration of  $PM_{2.5}$  and  $PM_{10}$  in the cave is also affected by the number of tourists in the peak tourist season in June–September 2015. In Table 1, the large temperature difference between outside and inside the cave is conducive to the migration of atmospheric particles. The higher the temperature, the stronger the airflow activity and the easier the diffusion of pollutants. When the relative humidity inside the cave is high, the particles are easily adsorbed on the surface of water vapor, which forms wet deposition, and then settles to the ground (Wang and Ogawa, 2015), thus reducing the concentration of suspended particles in the air. Precipitation in summer and autumn has a strong scavenging effect on pollutants (Tai et al., 2010). Concentration of particulate matter outside the cave is more likely to be affected by precipitation, as increase in humidity in the cave may cause the occurrence of salt damage to the murals. During the period of no air pollution, opening the cave door and increasing the air circulation plays an important role in reducing the concentration of suspended particulate matter in the air of the cave. Under high pressure, particulate matter does not readily diffuse, which easily causes  $PM_{2.5}$  and  $PM_{10}$  to accumulate (Plocoste, 2022). High wind speed is conducive to the migration of particulate matter in the air. The main reason may be the abundant deposits in the Mingsha Mountain and the gravel Gobi area at the top of the cave. The average annual wind speed at the top of the cave is  $4.23 \text{ m/s}$ , and the frequency of sand-blowing wind ( $5 \text{ m/s}$ ,  $2 \text{ m}$  height) is as high as 69.80%. Higher wind speed can easily blow the pollutants deposited on the sand dunes in the Gobi area into the cave (Tan et al., 2016). The westerly wind (strong with the low frequency) is harmful to the murals due to sand transportation and accumulation in front of the cave (Li et al., 2012). The topography of the valley between the Sand Mountain and Sanwei Mountain to the east is not conducive to the diffusion of pollutants. The average annual wind speed in the cave is  $0.54 \text{ m/s}$ , and the frequency of sand blowing is only 0.01%, which is not conducive to pollution. The higher the wind speed is, the easier the particles infiltrate into the caves from the cracks of Cave 16, increasing the mass concentration of particles in the cave. On the other hand, the disturbance of tourists causes the secondary suspension of particles.

#### 4.2 Sources of $PM_{2.5}$ and $PM_{10}$

Studies have found that soil in the top 50 cm in the Gobi area has high salinity (4.4% on average), and mainly consists of  $\text{Na}_2\text{SO}_4$  and  $\text{NaCl}$  (Li et al., 2017). As shown in Table 2, salt content of the Gobi area in the Mogao Grottoes may affect the content of inorganic ions ( $\text{Na}^+$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$ ) in atmospheric particulates, and effective control of unstable surface that is prone to dust can reduce the sources of harmful salt ions in the murals. The concentration of  $\text{NH}_4^+$  in the fine particles inside the cave was relatively high. The easier it is for fine particles to enter the fresco cracks, the greater the contribution of enriched salt ions to the mural degradation. Therefore, when removing dust on the surface of pigmented layer, more attention should be paid to the removal of fine particles.  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  mainly comes from secondary pollution, such as coal combustion and soil dust in the surrounding rural areas,  $\text{NO}_3^-/\text{SO}_4^{2-}$  can distinguish mobile pollution sources and fixed pollution sources (Yao et al., 2002). In this study,  $\text{NO}_3^-/\text{SO}_4^{2-}$  in  $PM_{2.5}$  and  $PM_{10}$  inside and outside the cave was less than 1, suggesting that coal combustion or dust contributes more



pollutants to the Mogao Grottoes than mobile source pollution. This is because motor vehicles are strictly prohibited in key protected areas, and tourist shuttle bus parks in an open area more than 200 m away from the cave, and buffering effect of protective forest belt in front of the cave effectively reduces the emission of secondary pollutants. This result is consistent with Yang et al. (2021), who found the significant impact of tourist visiting patterns on water-soluble ions in the Mogao Grottoes. In Figure 5a and b, May–August 2015 is the peak tourist season for the Mogao Grottoes, and tourists at this time account for more than 80% of the annual total.  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  may come from building sites and/or soil dust and dust weather in northern China (Li et al., 2015b), coal burning for heating, dust and pollutants from bare ground in winter, leading to increasing concentrations of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , and  $\text{Ca}^{2+}$ .  $\text{K}^+$  is an important indicator of dust and biomass burning (Li et al., 2003), the concentration of  $\text{K}^+$  was also relatively high from October to December 2015, and mainly derived from the burning of crop stalks such as corn and cotton in rural areas of Dunhuang.  $\text{F}^-$  mainly comes from ceramics, coal burning, and so on (Chen et al., 2013), the concentration of  $\text{F}^-$  was higher in July–August 2015, which may be mainly affected by the secondary suspension of particles containing coal combustion products caused by disturbance from tourists. The high concentration of  $\text{NO}_3^-$  from November 2015 to January 2016 may be related to the burning of coal in rural areas of Dunhuang for heating as well as sandy weather. The high concentration of  $\text{NH}_4^+$  from October 2015 to January 2016 was mainly related to the burning of coal in rural areas of Dunhuang. In Figure 5c and d, it may be that the lack of wind and more rainfall in June–September reduced water-soluble ions concentration outside the cave. In January–March 2016, cold winter air contained fugitive dust from bare ground pollutants and rural heating coal, impacted the water-soluble ions.  $\text{F}^-$  occurred from January to March 2016, may be mainly affected by the long-distance transmission of crop straw combustion products (Chen et al., 2013). The high concentration of  $\text{NH}_4^+$  from October 2015 to March 2016 was mainly related to the burning of coal for heating in the rural areas of Dunhuang. The content and type of water-soluble ions in the semi-closed caves are mainly affected by air exchange and tourist activities, while those outside the caves are mainly affected by natural environment. Therefore, in the period of frequent sandstorms, numerous tourists and vehicles, intensive crop burning and coal heating, the air exchange inside and outside the cave can be reduced by replacing the existing cave door. When the pollutant level is high, the internal air purification equipment and other measures can effectively reduce the concentration of water-soluble ions inside the cave.

#### 4.3 Change of water-soluble ion in relation to dust storm

Change of particulate and water soluble ions is closely related to the extreme process of sand and duststorm in this area. Some research has shown that the concentration of particles, especially  $\text{PM}_{10}$ , during duststorm is much higher than that during non-duststorm (Guan et al., 2017), and chemical elements inside a cave are affected by weather conditions outside the cave (Whittlestone et al., 2003), including precipitation, clear or cloudy conditions and wind. Furthermore, the physical force generated by the large sand particles during a strong duststorm causes scratches on the surface of murals (Ogura et al., 2021). This study found that duststorm and precipitation have greater impact on pollutants inside and outside of the cave. That is why closing the cave and stopping visitors from visiting reduce the transmission of pollutants from outside to inside the cave. The effects of duststorm and precipitation on water-soluble ions are similar to those of Xu et al. (2020) and He et al. (2020). The sand and saline soil in the area deposited by the Daquan River, Kumtag Desert and Gobi area, and redistributed by wind influenced the water-soluble ions in atmospheric particles (Zhang et al., 2007).

Disruption and blistering of murals were mainly caused by  $\text{NaCl}$  and  $\text{Na}_2\text{SO}_4$  (Jin et al., 2016). If particulate matter containing a large amount of  $\text{Na}^+$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$  is deposited on or attached to wall paintings, and the humidity in the cave reaches more than 65%, the salt in the ground and the murals will be activated, resulting in the dissolution and crystallization of soluble salts and accelerating mural damage, such as blistering (Demas et al., 2015; Li et al., 2015a). During the

duststorm, the air mass mainly originated from the Hami area of Xinjiang, and was transported to the Dunhuang through the Taklimakan Desert and Kumtag Desert. This process may cause long-distance transport of atmospheric particles or short-distance local dust formation in the surrounding deserts and Gobi area. Therefore, during duststorm and sudden heavy rainfall, we recommend closing the caves to reduce the inward transmission of pollutants and humid air. At the same time, after duststorm, the dust accumulation in the caves and on the doors, windows, floors, and other areas should be cleaned up in time. The results of this study can be used as a reference for the improvement of early warning system of duststorm environment in the Mogao Grottoes and the decision-making of opening of the cave in response to extreme weather in advance. Deposition of alkaline or acidic particles on the relics will cause potential or immediate acid and salt corrosion of the surface and the color in a high-humidity environment (Mašková et al., 2017). In this study, the results showed alkalinity in the Mogao Grottoes, consistent with those at the Jinsha Site Museum (Deng et al., 2022), but contrary to the results of the Hanyang Tomb (Jia et al., 2015). The climate of Dunhuang is extremely arid, with low precipitation and high evaporation, and the salinity stays on the surface as evaporated groundwater makes the soil alkaline. There are no industrial pollution sources in the protected area. The alkalinity of atmospheric particulate matter in the Mogao Grottoes may be related to factors such as more crustal elements present in the cave and low levels of car exhaust pollution.

Water-soluble ion pollution has complex origins comprising sea fog emissions, soil dust, biomass combustion, and fossil fuel combustion emissions (Wang, 2005). In Tables 3 and 4,  $\text{Na}^+$  and  $\text{Cl}^-$  may come from Gobi area and the Daquan River, and  $\text{NH}_4^+$  is mainly derived from secondary pollution sources such as rural coal combustion.  $\text{K}^+$  derived mainly from the crop straw burning. In summary, the pollution sources of atmospheric particulates of different particle sizes in the Mogao Grottoes are diverse and include mainly straw burning, secondary pollution sources, soil dust, dry spring rivers, and from tourist activities. Therefore, it is necessary to strengthen the control of the burning of agricultural crops in Dunhuang, reduce the operation of motor vehicles (especially private cars of employees) in protection zone of the Mogao Grottoes, increase the popularization of new energy commuter vehicles, and reduce the cracking sand hills and dry springs on the top of the caves.

## 5 Conclusions

Concentrations of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  inside and outside the cave were the highest in March 2016 and the lowest in December 2015. The lower particulate matter concentration from June to September is associated with good diffusion conditions and increased precipitation and reduced wind impact.  $\text{Ca}^{2+}$ ,  $\text{NH}_4^+$ ,  $\text{Na}^+$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$  were main components. There was a consistent relationship of  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  between the inside and outside of the cave, and the particulate source was mainly coal combustion or dust from the ground, which contributed more pollutants in the Mogao Grottoes, with less mobile source pollution. Duststorm can lead to sharp increases in the air particulate concentration and can cause predictable deterioration of mural covers. The control of atmospheric particulate matter pollution in the Mogao Grottoes should be based on improving the comprehensive system of wind and sand prevention and control, limiting the number of tourists, adjusting and opening the caves, and cleaning the ground of sand or dust deposits in a time, as well as adding effective ventilation and air purification and other intelligent control systems.

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